

Section 9 1 Carbon Compounds Answers

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organic compounds. 262 Chapter 9 FOCUS Objectives 9.1.1 Relate the structures of three forms of carbon to their properties. 9.1.2 Explain why there are millions of different organic compounds. 9.1.3 Relate the number and arrangement of carbon atoms in hydrocarbons to their properties. 9.1.4 Distinguish unsaturated from saturated hydrocarbons. 9.1.5 Classify hydrocarbons using structural formulas and names. 9.1.6 Describe the formation,

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Most of the compounds are hydrocarbons, molecules made up of only hydrogen and carbon atoms only. There are different forms of hydrocarbons but the most common type are known as alkanes. Alkanes are a homologous series; they have the similar properties and follow a certain pattern of chemical formula.

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organic compounds. 262 Chapter 9 FOCUS Objectives 9.1.1 Relate the structures of three forms of carbon to their properties. 9.1.2 Explain why there are millions of different organic compounds. 9.1.3 Relate the number and arrangement of carbon atoms in hydrocarbons to their properties. 9.1.4 Distinguish unsaturated from saturated hydrocarbons. 9.1.5 Classify hydrocarbons using structural formulas and names. 9.1.6 Describe the formation, composition, and uses of

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9.1 Carbon Compounds. STUDY. PLAY. the two elements that all organic compounds contain? carbon and hydrogen. ... section 9.1 Carbon Compounds. 20 terms. Science. 80 terms. AQA C1 Keywords and definitions. 18 terms. Hydrocarbons. OTHER SETS BY THIS CREATOR. 46 terms. Civilization (Unit 1) 11 terms. 15.3.

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Section 9.1 9.1 Carbon Compounds All organic compounds contain hydrogen, carbon, and oxygen. Answers are: Carbohydrates are made of carbon, hydrogen, and oxygen.

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Liquid carbon disulfide reacts with oxygen gas, producing carbon dioxide gas and sulfur dioxide gas. CS 2(l) 3O 2(g) 0 CO 2(g) 2SO 2(g) 6. Challenge A piece of zinc metal is added to a solution of hydrogen sulfate. This reaction produces a gas and a solution of zinc sulfate. Zn(s) H 2SO 4(aq) 0 H 2(g) ZnSO 4(aq) Section 9.1 Assessment page 288 7.

[Chemical Reactions](#)Chemical Reactions

SECTION 9.2 NAMING AND WRITING FORMULAS FOR IONIC COMPOUNDS (pages 260 – 266) This section explains the rules for naming and writing formulas for binary ionic compounds and compounds containing a polyatomic ion. Binary Ionic Compounds (pages 260 – 263) 1. Traditionally, common names were based on some _____ of

[Name Date Class CHEMICAL NAMES AND FORMULAS 9](#)

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Carbon Compounds Study Guide Answers Organic compounds contain a. carbon and usually other elements. The number of covalent bonds a carbon atom can form with other atoms is c. 4. Biology Section 3-1 Review: Carbon Compounds Flashcards ... Start studying section 9.1 Carbon Compounds.

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1 Only living things can make carbon compounds. 2 Hydrogen atoms often bond with carbon to form compounds. 3 Simple sugars are the building blocks of proteins. 4 Carbon atoms can form single, double, and triple covalent bonds. 5 The suffix in the name of an organic compound indicates the kind of bonds joining the carbon atoms.

[Carbon Chemistry](#)

David Cameron, in Supramolecular Photosensitive and Electroactive Materials, 2001. 1. INTRODUCTION. Nitrogen-containing organic and polymeric carbon compounds have been known for many years, and even compounds containing only carbon and nitrogen, for example, so-called paracyanogen compounds that have a polymerized (C=N) n structure have been known and have been synthesized for some time [1].

[Carbon Compounds - an overview | ScienceDirect Topics](#)

9. C 10. carbonic acid 11. carbon, oxygen, and hydrogen 12. chemical change 13. D 14. aluminum oxide SECTION: MIXTURES 1. mixture 2. mixture 3. compound 4. Answers will vary. Sample answer: You can see each component in the pizza. Each component has the same chemical makeup as it did before the pizza was made. 5. physical 6. A 7. B 8. D 9. C 10 ...

[Skills Worksheet Directed Reading A](#)

3 Chemical and Physical Properties Expand this section. 4 Related Records Expand this section. ... 4.1 Related Compounds with Annotation. Help. New Window. PubChem. 4.2 Component Compounds. Help. New Window. ... 9.1.4 NORMAN Suspect List Exchange Classification. Help. New Window.

[Nickle carbonyl | C4NiO4 - PubChem](#)

CHAPTER 1 ANSWERS AND SOLUTIONS TO ODD-NUMBERED PROBLEMS Carbon, The Element of Organic Compounds (Section 1.1) 1.1 Why were the compounds of carbon originally called organic compounds? Solution The term, organic compound, originally referred to those compounds found in living, " organic " matter. 1.3 Describe what W ö hler did that made the vital force theory highly questionable.

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