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Ethanol (CH3CH2OH) is being used as a gasoline additive or alternative in many parts of the world.(a) Use bond energies to find ?H°rxn for the combustion of gaseous ethanol. (Assume H2O forms as a gas.)(b) In its standard state at 25°C, ethanol is a liquid. Its vaporization requires 40.5 kJ/mol. Correct the value from

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