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acids and bases: Acid and Base reactants nitric acid and sodium hydroxide hydroiodic acid and calcium hydroxide magnesium hydroxide and hydrosulfuric acid ammonium hydroxide and hydrofluoric acid barium hydroxide z and sulfuric acid chloric acid and rubidium hydroxide Q) calcium hydroxide (C 2 and carbonic acid CD Name of ionic compound Formula

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Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water.

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