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a chemical change in which one element replaces a second element in a compound; displacement reaction Activity Series a list of elements in order of decesasing activity

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10.12 Simultaneous and Sequential Chemical Reactions Chapter 11 States of Matter 11.1 Factors That Determine Physical State 11.2 Property Differences among Physical States 11.3 The Kinetic Molecular Theory of Matter 11.4 The Solid State 11.5 The Liquid State 11.6 The Gaseous State 11.7 A Comparison of Solids, Liquids, and Gases

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• A skeleton equation is a chemical equation that does not indicate the relative amounts of the reactants and products. Like a recipe with no amounts listed.

Chapter 11

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11.2 Types of Chemical Reactions> 13 A decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or more products. • The products can be any combination of elements and compounds. • Most decomposition reactions require energy