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$\text{Cu}(\text{NO}_3)_2 + \text{Zn} \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Cu}$  There is a sentence that you can \$,rite to describe what happens above: Copper1D nitrate reacts withzinc metal and yields zinc nitrate and copper metal. Take a minute to notice the relationship between the sentence and the symbols in the reaction

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ChemQuest 23 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Drawing ... Hint: add your answer to

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question 1 and four times your answer to question 2.  $4 + 4(7) = 4 + 28 = 32$  4. Verify that there are 32 electrons pictured in the drawing of  $\text{CCl}_4$ . Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line represents two electrons). 5 ...

ChemQuest 23 : Drawing Covalent Compounds

Page 37 and 38: 37 ChemQuest 12 Name: \_\_\_\_\_ Page 39 and 40: 39 Critical Thinking Questions 6. H; Page ... 143 9. Consider your answers to que. Page 145 and 146: 145 Critical Thinking Questions 5. Page 147 and 148: 147 8. If a substance is a Bronsted. Page 149 and 150: 149 6. Use the information given an . Page 151 and 152: 151 ChemQuest 50 Name: \_\_\_\_\_ Page 153 and 154: 153 11. Find the pH of a ...

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